

Lipids (1. LF UK, NT)

They are natural non-polar compounds that are insoluble in water, but are soluble in non-polar solvents.

Importance of lipids

- they are one of the basic components of food, they serve as a source and reserve of energy (38kJ/g)
- structural function - they are part of biomembranes (eg: double layer of phospholipids -> form micelles)
- protective function - they cover some organs and thus protect them from shock
- thermal insulation
- solvents of non-polar vitamins (A, D, E, K)
- precursors

Divisions

Simple

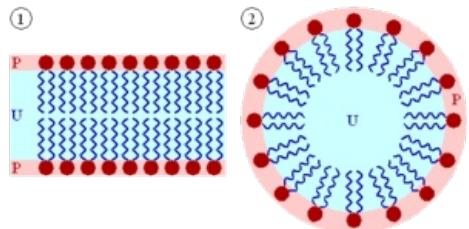
they contain only the lipid part

- acylglycerols - are esters of VMK and glycerol (fats and oils)
- triacylglycerols are the most abundant in the human diet
- waxes (animal, vegetable)

Composite

obsahují lipidovou a nelipidovou část

- they contain a lipid and a non-lipid part
 - glycoacylglycerols - contain a carbohydrate component
- are part of plant membranes
- phosphoacylglycerols - are part of biological membranes and lipoproteins
- their molecule is amphipathic
- sphingolipids - their basis is a compound that contains the 18-carbon amino alcohol sphingosine
- they are further divided into sphingomyelins containing ceramide and choline, these include cerebrosides and gangliosides containing carbohydrate and sialic acid - they are amphipathic and are found in the brain and nerves



Derived

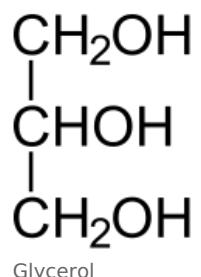
- terpenes
- steroids

Higher fatty acids

They are the basic component of lipids. They have a high number of C, a high number of non-polar bonds and only one functional group - COOH, which is polar. They are unbranched, have a hydrophobic character and a cis arrangement that is natural in nature and our body can break it down.

Glycerol

=propane-1,2,3-triol, glycerin
It is a sweet viscous liquid, infinitely miscible with water.



Source of dietary fat

extraction of crude fats and oils, sources

- **vegetable** - pressing, extraction
- **animal** - smelting, extraction

Vegetable fats and oils

Refining

- slime removal (hydration), vegetable slimes, proteins, their complexes - lecithin
- deacidification (neutralization) - acid salts
- whitening - carotenoids , chlorophylls
- deodorization - tocopherols , sterols

Classification

By consistency

- oils (liquid)
 - drying - linen
 - semi-drying - sunflower/soybean
 - non-drying - olive
- fats (plastic, mushy) - lard
- waxes (hard, non-greasy) - beeswax

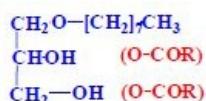
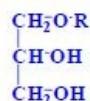
By structure

1. fatty acids and their soaps R-[CH₂]_n-COOH
2. homolipids (esters of fatty acids with alcohols)

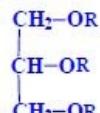
1. monohydric alcohols (waxes)

- aliphatic (cerides)
 - CH₃-[CH₂]₂₅-OH... ceryl alcohol (beeswax)
 - H₃-[CH₂]₁₅-OH... hexadecan-1-ol, cetyl alcohol (cetaceum)
- alicyclic (steroids) - esters of sterols (cholesterol), triterpene alcohols

2. dihydric alcohols (glycols), Alkoxylipids: 1-Alkoxypropane-2,3-diols Chimyl alcohol



3. trihydric (glycerol) fats and oils

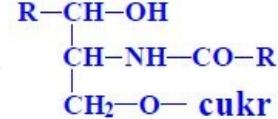
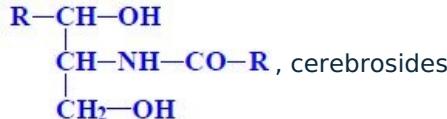
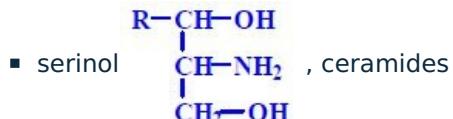
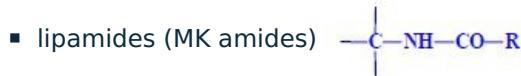
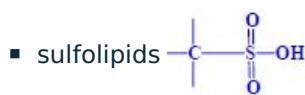
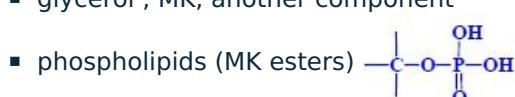


4. polyhydric alcohols:

- sugars - glycolipids
 - galactose, galactosides, digalactosides, diacylglycerogalactoside (diacylgalactosylglycerol)
$$\begin{array}{c} \text{CH}_2\text{O-COR} \\ | \\ \text{CHO-COR} \\ | \\ \text{CH}_2\text{O-galaktosa} \end{array}$$
 - sacrose (1-3 MK emulsifiers, 6-8 MK low-energy fats (OLESTRA))
 - sorbitol (sugar alcohols) emulsifiers

3. heterolipids 0.5-2%

- glycerol , MK, another component



4. complex lipids

- proteolipids (lipoproteins)

- glycolipids (cerebrosides)
- mucolipids (sialoglycosphingolipids = gangliosides)

Links

Related articles

- Fatty acid
- Lipid breakdown and metabolism of ketone bodies

References

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- DAVÍDEK, Jiří. 3. LIPIDY [online]. [cit. 2012-03-11]. <<https://el.lf1.cuni.cz/p60846053>>.